1. (Currently amended) A process for preparing a from 5 to 60% by weight aqueous alkaline solution of a reduced indigoid dye, which comprises reducing said indigoid dye electrochemically in the presence of a mediator, where the dye to be reduced is not precharged all at once, but is added a little at a time.

2. (Original) A process as claimed in claim 1, wherein said mediator is an iron (II/III) complex salt.

3-8 cancelled

9. (Previously presented) The process as claimed in claim 1, wherein from 0.003 to 0.08 mol of mediator is used per mole of dye.

10. Cancelled

11. (Previously presented) The process as claimed in claim 1, wherein the alkali used is a mixture of at least two alkali metal hydroxides wherein none of the alkali metal hydroxides accounts for more than 70 mol%.

12. (Previously presented) The process as claimed in claim 1, wherein from 1.2 to 2 mol of alkali are used per mole of dye.

13. (Previously presented) The process as claimed in claim 1, wherein said reducing is effected at from 10 to 80°C.

14. (Previously presented) The process as claimed in claim 1, wherein the solution prepared is a solution from 15 to 45% strength by weight leuco indigo solution.

15. (Previously presented) The process as claimed in claim 2, wherein from 0.003 to 0.08 mol of mediator is used per mole of dye.

16. (Previously presented) The process as claimed in claim 15, wherein the alkali used is a mixture of at least two alkali metal hydroxides wherein none of the alkali metal hydroxides accounts for more than 70 mol%.

17. (Previously presented) The process as claimed in claim 16, wherein from 1.2 to 2 mol of alkali are used per mole of dye and said reducing is effected at from 10 to 80°C.

18. (Previously presented) The process as claimed in claim 17, wherein from 0.008 to 0.05 mol of mediator is used per mole of dye.